

19. Nitrogen compounds

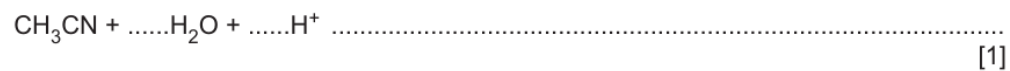
19.1 Primary amines

Paper 2

Question Paper

1 (b) H_2O reacts with both inorganic and organic compounds.

(ii) Write an equation for the reaction of CH_3CN with H_2O in acidic conditions.



2 Fig. 3.1 describes a sequence of reactions that can be used to produce a food additive, compound **Y**, from $\text{CH}_3\text{CH}_2\text{Cl}$.

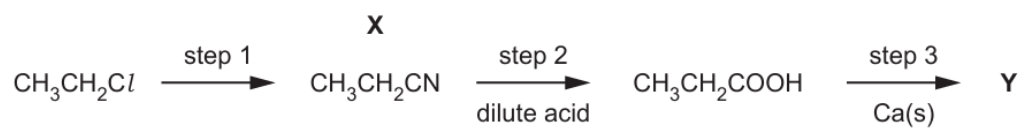


Fig. 3.1

(a) (iii) Identify the type of reaction that occurs when dilute acid is added to **X** in step 2.

..... [1]

- 3 (c) $\text{Ba}(\text{OH})_2$ is used to hydrolyse organic compounds.

Fig. 2.2 shows the reaction of **B** with $\text{Ba}(\text{OH})_2$, followed by acidification.

Draw the structures of the organic products of the process shown in Fig. 2.2.

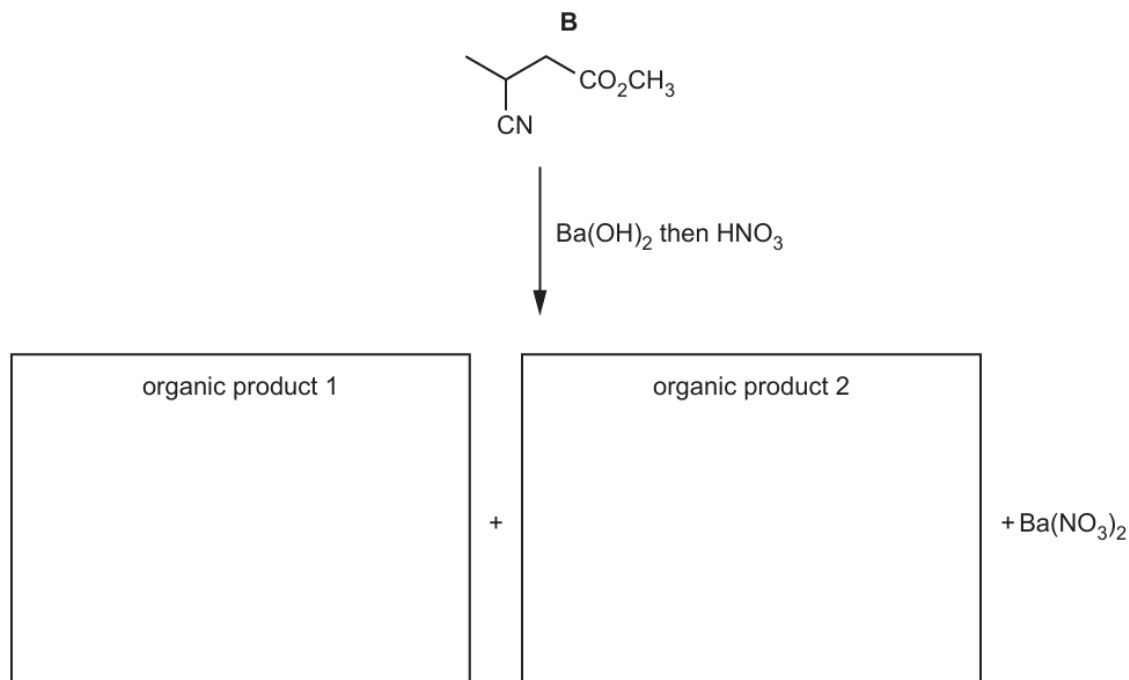


Fig. 2.2

[3]

- 4 **S** is a secondary alcohol with molecular formula $\text{C}_4\text{H}_{10}\text{O}$.

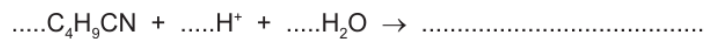
(b) **S** is converted to **V** in a three-step reaction sequence.



In step 1, the secondary alcohol **S** reacts with PBr_3 to produce **T**, which has molecular formula $\text{C}_4\text{H}_9\text{Br}$.

(iv) Step 3 involves heating $\text{C}_4\text{H}_9\text{CN}$ with dilute acid to form **V**.

Complete the equation for this reaction.



[2]